

The structure of an atom.

1. Noun
2. Verb
3. Preposition
4. Adjective
5. Verb
6. Noun
7. Article
8. Verb
9. Noun
10. Noun
11. Verb
12. Noun
13. Noun
14. Noun

The structure of an atom.

An ______{Noun} is so small that billions of them could ______{Verb} on the head of a pin. ______{Preposition} atoms are made up of even ______{Adjective} particles called neutrons, protons, and electrons. Neutrons and protons are ______{Verb} at the center of the ______{Noun} which is called ______{Article} nucleus. Protons are positively ______{Verb} particles, and neutrons are particles that have no charge. Electrons are negatively charged particles that are located outside the ______{Noun}. ______{Noun} constantly ______{Verb} around an atom's nucleus in energy levels. The basic structure of an ______{Noun} is the result of the attraction between ______{Noun} and ______{Noun}. Atoms contain an equal number of protons and electrons, so the overall charge of an atom is zero.