

The structure of an atom

1. Noun
2. Number
3. Noun
4. Noun
5. Verb
6. Plural Noun
7. Plural Noun
8. Noun
9. Adjective
10. Adjective
11. Noun
12. Noun
13. Verb
14. Adjective
15. Noun
16. Noun
17. Noun
18. Noun
19. Number

The structure of an atom

An _____ Noun _____ is so small that _____ Number _____ of them could fit on the head of a pin. Yet, _____ Noun _____ are made up of even smaller particles called _____ Noun _____ protons, and electrons, as _____ Verb _____ in figure 1.

_____ Plural noun _____ and _____ Plural noun _____ are located at the center of the atom, which is called the _____ Noun _____.

Protons are _____ Adjective _____ charged particles and neutrons are particles that have no charge. Electrons are

_____ Adjective _____ charged particles that are located outside the _____ Noun _____. _____ Noun _____ constantly move

_____ Verb _____ an atom's nucleus in energy levels. The basic structure of an atom is the result of the

_____ Adjective _____ between _____ Noun _____ and _____ Noun _____. _____ Noun _____ contain an equal number of protons and electrons, so the overall charge of an _____ Noun _____ is _____ Number _____.