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The structure of an atom

1. Noun _____
2. Number _____
3. Noun _____
4. Noun _____
5. Verb _____
6. Plural Noun _____
7. Plural Noun _____
8. Noun _____
9. Adjective _____
10. Adjective _____
11. Noun _____
12. Noun _____
13. Verb _____
14. Adjective _____
15. Noun _____
16. Noun _____
17. Noun _____
18. Noun _____
19. Number _____

The structure of an atom

An _____Noun_____is so small that _____Number_____of them could fit on the head of a pin.Yet, _____Noun_____are made up nof even smaller particless called _____Noun_____ protons, and electrons, as _____Verb_____in figure 1. _____Plural noun_____and _____Plural noun_____are located at the center of the atom, which is called the _____Noun_____. Protons are _____Adjective_____charged particles and neutrons are particles that have no charge. Electrons are _____Adjective_____charged particles that are located outside the _____Noun_____. _____Noun_____constantly move _____Verb_____an atom's neucleus in energy levels. The basic structure of an atom is the result of the _____Adjective_____between _____Noun_____and _____Noun_____. _____Noun_____contain an equal number of protons and electrons, so the overall charge of an _____Noun_____is _____Number_____.

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